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SAPONINS AND SAPOGENINS XXV^{*} - THE SAPOGENIN OF ACACIA CONCINNA D.C. PODS AND THE CONSTITUTION OF ACACIC ACID

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The pods of Acacia concinna D.C. by alcoholic extraction yields a colourless saponin, which on hydrolysis with sulphuric acid yields an acid genin, m.p.268-72°, diacetyl lactone m.p.235- 36° . This acid has been identified as acacic acid (I)¹ by mixed melting point with authentic samples of acacic acid and its acetyl lactone respectively, (Cf.Acacic acid m.p.275-76°; acacic acid acetyl lactone m.p.235-36°)¹.

Acacic acid is a sapogenin of fairly common occurrence. Apart from the bark of Acacia intsia¹, it has also been reported in the bark of Albizzia lebbek² and in the seeds of Acacia intsia³, Acacia concinna³, Albizzia stipulata⁴ and Albizzia odoratissima⁵. The preliminary studies fixed acacic acid as a trihydroxy monocarboxylic triterpenic acid of either tetracyclic

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2055

er α -amyrin group^{*}. Later, one of the authors^{6,7} on biogenetic grounds favoured a β -amyrin type structure and also proposed 3,16,21 trihydroxy-olean-12-ene-28-oic acid (I) structure for acacic acid. This paper describes the chemical evidences which support this formulation (I).

Acacic acid acetyl lactone (II) on treatment with methanolic hydrochloric acid (10%) on a water bath, gave a product m.p.257-61° (III). The product (III) on oxidation (with chromic acid and pyridine in cold) furnished a neutral acetyl diketone m.p. 338-42° (IV) (Found: C, 76.50; 76.44; H, 8.97; 9.13; Calod.for C₃₁H₄₆0₄: C, 77.13; H, 9.61) due to the formation of a β -ketonic acid which decarboxylates easily to yield a neutral diketone (IV), with the loss of one molecule of carbon dioxide. The ketone (IV) on Wolff-Kishner reduction again yielded a neutral product m.p.183-88 (V) (Found: C, 84-20; H, 11.73, Calcd.for C29H480: C, 84.40; H, 11.72). The product m.p.183-88 (V) on acetylation with acetic anhydride and pyridine in cold gave an acetate m.p. 210-16⁹ (VI). The products (V) and (VI) have been found to be identical with 28-nor β -amyrin (Nor-echinocystenol A) and its acetate respectively by mixed melting points with authentic samples obtained from echinocystic acid⁸.

^{*}Z.Naim Ph.D. Thesis, Aligarh Muslim University, Aligarh, 1962 suggested it to have an *construction* and two secondary hydroxyl groups and a carboxyl group.



The formation of 28-nor β -amyrin (V) from Acacic acid (I) proves beyond doubt that acacic acid is a member of the β -amyrin group with one of the hydroxyl groups at C-3, carboxyl group at 28 and the double bond between C-12 and 13. The position of the second hydroxyl group was taken to be at C-21 due to the formation of a Y-lactone (II) in the acetylation process. Finally the position of the remaining i.e. the third hydroxyl group was indicated at C-16 or C-22 by the fact that the carboxyl group easily decarboxylates on oxidation due to the formation of a β -ketonic acid. Out of these two positions the third hydroxyl was finally fixed at C-16 as acacic acid on periodic acid exidation was recovered unchanged thereby eliminating position C-22 as it is adjacent to the 21-hydroxyl group. Therefore the constitution of acacic acid has finally been fixed as 3,16,21-trihydroxy olean-12-ene-28-cic acid (I).

Further work on the stereochemistry and conforma-tion is in progress.

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